Synthesis of D-Galactosamine and D-Allosamine Derivatives via a Microwave-Assisted Preparation of 1,6-Anhydroglucosamine

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Supporting Information

ABSTRACT: We report a microwave-assisted intramolecular anomeric protection (iMAP) of glucosamine, which facilitates concise transformation of 1,6-anhydroglucosamine into 1,6anhydrogalactosamine and 1,6-anhydroallosamine. The iMAP simultaneously obviates both the O1 and O6 protection, and the differentiation between O3 and O4 can be well-controlled by the N2 functionality because of the hydrogen bonding between N2



and O4. Epimerization of O4 afforded the galactosamine derivative and that of O3 yielded allosamine.

2-Acetamido-2-deoxy-D-glucose (D-GlcNAc) and 2-acetamido-2-deoxy-D-galactose (D-GalNAc) are the most abundant types of 2-amino-2-deoxyhexose in nature.¹ They are found in oligosaccharides and glycoconjugates, such as glycolipids, peptidoglycan, glycoproteins, and glycosaminoglycans.² Their *N*-acetylated and sulfonated derivatives are widely distributed in type A blood group antigen, GPI anchors, and lipopolysaccharides. D-GlcNAc is inexpensive and can be commercially purchased in large quantities.³ However, most other 2-amino-2deoxyhexoses, such as allosamine and galactosamine, are rare and much more expensive. Therefore, economically viable and efficient strategies must be developed for synthesizing these sugars.

Galactosamine and its derivatives are the building blocks for synthesizing numerous biologically essential carbohydrate molecules. For example, N-acetylated derivatives of galactosamine are present in chondroitin sulfate and Tn antigen. Bacterial cell walls contain large amounts of capsular polysaccharides (CPS) and lipopolysaccharides (LPS), which are responsible for shock absorbance and biosynthesis. These components are crucial virulence factors and promote bacterial colonization, block phagocytosis, and interfere with leukocyte migration and adhesion. Amino sugars are essential components of the bacterial glycoconjugate, responsible for hostpathogen interaction.³ Furthermore, N-acetylated derivatives of allosamine are present in natural products. For example, allosamine-an insect Chitinase inhibitor isolated from the mycelium of Streptomyces sp-was detected in allosamidin in 1986,⁴ and the allosamine core is an indispensible component in streptothricin antibiotic.4

Hydrolysis of chondroitin sulfate under acidic condition is the major source of D-galactosamine.⁵ Several methods have

been reported for the synthesis of D-galactosamine, including the extension of D-lyxose chain, epimerization of C4 of Dglucosamine, \int_{-1}^{0} addition of ammonia to 1,6:2,3-dianhydro- β -Dtalopyranose.⁷ and azidonitration.⁸ The addition of nitrosyl chloride to tri-O-acetyl-D-galctal followed by the reduction of oxime also yields galactosamine.⁹ However, in most of these cases, the yields were low, and a high degree of stereoselectivity cannot be achieved. Recently, Kulkarni et al. reported the rapid transformation of D-mannose into protected D-galactosamine and D-glucosamine thioglycosides.¹⁰ Allosamine is a C3 epimer of glucosamine. This rare amino sugar is not commercially available; it exerts insecticidal activity through the inhibition of ecdysis.⁴ Various mechanisms have been reported for the synthesis of this rare sugar, including the elongation of ribose chain,¹¹ hydrolysis of oxazoline,¹² inversion at *C*-3 via the Mitsunobu approach,¹³ diazotransfer followed by inversion of O3,¹⁴ and stereoselective formation of imine followed by reduction.¹⁵ However, these approaches result in low stereoselectivity and poor yields.

The preparation of sugar building blocks generally entails laborious anomeric protection followed by the traditional protection and deprotection strategies. Short, efficient, and improved methods for the synthesis of these potent sugars are desired. We adopted the intramolecular anomeric protection (iMAP), an efficient strategy 1,6-cyclization of free sugar. Through our iMAP method, excellent yields of these sugars can be generated in a short time by using only catalytic amount of Lewis acid¹⁶ with no additional or excessive reagents

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required;¹⁶ therefore, the traditional anomeric and primary *O6* protection can be simultaneously obviated without forming the α/β anomer mixtures; protection–deprotection sequence can therefore be greatly reduced. Besides, the configuration of the hydroxyl groups is altered from equatorial to axial after the 1,6-anhydro ring formation, thus altering the reactivity. Therefore, several factors can be used to control the regioselectivities among the remaining secondary hydroxyl groups. Here, we utilize iMAP for the synthesis of galactosamine and rare allosamine through the regioselective protection of 1,6-anhydroglucosamine.

N-functionalization of amino sugars is often tedious. To simplify the protocol, we recently reported a simple and efficient protocol for the chemoselective per-*O*-trimethylsilylation of amino sugars followed by *N*-functionalization.^{17,18} Recently, we developed a microwave-assisted method for the synthesis of 1,6-anhydro sugars from free sugars using TMSOTf and TfOH as a catalyst in moderate to good yields.¹⁶ We applied a similar protocol for the synthesis of 1,6-anhydroglucosamine. The silyl derivative of glucosamine was subjected to microwave irradiation to afford 1,6 cyclization **5**, **6**, 7 in 85%, 70%, and 70% yields, respectively (Scheme 1). The

Scheme 1. Microwave-Assisted Synthesis of 1,6-Anhydroglucosamine



structure of the bicyclo skeleton was clearly observed in a single X-ray crystal structure (Figure 1). The rigid conformation



Figure 1. ORTEP diagram of 7 with the thermal ellipsoids drawn at 50% probability level.

[3.2.1] bicyclo skeleton of 1,6-anhydroglucosamine derivatives 7 greatly simplifies the complexity of the glucosamine molecules by eliminating the problems of anomers and simultaneously obviating two protecting groups on the O1 and O6. In addition, the axial orientation of the equatorial bonds projects *N*HTCA and O4 in the same phase. We propose that the regioselective protection can be controlled by the amine functionality caused by the hydrogen bonding between N and O4 oxygen atoms.

X-ray crystallographic analysis (Figure 1) revealed that the distance between $O4 \rightarrow C2$ amide proton is 2.325 Å and $O4 \rightarrow C2$ nitrogen atom is 2.899 Å, indicating a strong hydrogen bonding. Thus, O4 can be expected to be less nucleophilic. Moreover, the rigidity of the sugar ring is expected to increase, and the steric hindrance can be used as an accurate indicator to

differentiate O3 and O4. Utilizing these factors facilitates the differentiation of O3 and O4 for the efficient preparation of other amino sugars. We estimated that the amine functionalities with an N-H bond, such as TCA or Troc group, would favor the protection at O3, but those without the N-H bond, such as azido group, would likely favor the O4 because of the steric hindrance between O3 and C6–O6 bond.

To synthesize galactosamine derivatives, our protocol involved the formation of 1,6-anhydrosugar followed by the reductive etherification at O3 and epimerization of O4. To differentiate O3 and O4, we performed the reductive etherification reaction of compound 7 using our established protocol.^{18,19} For the synthesis of bacterial glycoconjugate, we preferred 2-methylnaphthyl (2-NAP), a stable and versatile protecting group for the regioselective protection that can be easily removed using DDQ and CAN under mildly acidic conditions.^{20–22}

First, the reductive etherification of 7 with 2-naphthaldehyde and triethylsilane was examined at temperatures of -78 to 0 °C (Table 1). For the ease of characterization, the products were

Table 1. Optimization of the Regioselective Protection of 7

TMS TO O TMSO NHTCA	1. TMSOTf, MS 3 Å, 2-naphthaldehyde, Et ₃ SiH, <i>T</i> 0 °C, 5 h, dry DCM 2. TBAF, rt, 2 h, Ac ₂ O, Pyr., rt, 1 h	2-NAP 0 2-NAP 1 ACO NHTCA 2-NAPO 8	NHTCA 9
entry	T (°C)	8	9
1	-78	47	30
2	-50	50	32
3	-20	55	25
4	0	63	12
5	rt	42	23

treated with TBAF to cleave the trimethylsilyl group and then with pyridine and acetic anhydride; subsequently, the products were isolated as its acetylated derivatives. The reaction at -78°C provided only 47% of 8 and 30% of diaryl product 9 (Table 1, entry 1). A similar result was observed at -50 °C (Table 1, entry 2), whereas a slight improvement was observed when the reaction temperature was increased to -20 °C [55% of 8 (Table 1, entry 3)]. The reaction at 0 °C provided 63% of 8 and 12% of 9 (Table 1, entry 4). This result indicates that O3 is more reactive than O4.

The regioselective reductive etherification of 7 was further screened using various aryl aldehydes and the results were consistent in all cases, namely, **10**, **11**, **12** and **13** (Scheme 2). Regioselectivity was confirmed through X-ray crystallography of **11** (see Supporting Information).

These results briefly confirmed our hypothesis that the hydrogen bonding between the O4 and C2 N–H lowers the nucleophilicity of O4 and renders O3 more reactive despite the steric hindrance between O3 and C6–O6 bond. This protocol

Scheme 2. Regioselective Reductive Etherification of 7 by Using Various Aryl Aldehydes



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not only allows the synthesis of galactosamine but also provides the required building blocks for chain extension in oligosaccharide synthesis. The ORTEP of **11** indicates that despite the rigid [3.2.1] conformation after the 1,6-cyclization, in contrast to **11** (see Supporting Information), in the absence of the hydrogen bonding in **14**, the distance between $O4 \rightarrow N1$ of *C*2 azide elongated to 3.439 Å because of the 1,3-diaxial repulsion.

Without the hydrogen bonding that greatly deactivates the C4 hydroxyl group, O4 is more reactive than the sterically and conformationally more hindered O3 toward the regioselective reductive etherification (Scheme 3). The inductive effect of the

Scheme 3. Regioselective Reductive 2-Naphthylation of 5

	1. TMSOTf, MS 3 Å, 2-naphthaldehyde, Et₃SiH, -78 °C, 16 h, dry DCM		
TMSO N ₃	2. TBAF, 0 °C to rt, 2 h, Ac ₂ O, Pyr., rt, 2 h, 72%	2-NAPO N ₃	
5		14	

azide group must also enhance the regioselectivity by lowering the nucleophilicity of O3. To further verify this hypothesis, we repeated the reductive etherification by replacing the C2 NHTCA group with an azido group to 2-deoxy-2-azido-1,6anhydroglucosamine (5). In the absence of the hydrogen bonding, the regioselectivity was reversed when we used 5 as the substrate (Scheme 3). For easy NMR characterization, the product was again treated first with TBAF to remove the TMS group and then with pyridine and acetic anhydride to transform the product into its acetylated derivative. The C4 etherified compound 14 was isolated as a single regioisomer in 72% yield. The regioselectivity was reconfirmed through X-ray crystallography of 14 (see Supporting Information).

These results showed that the regioselective protections between the C3 and C4 hydroxyl groups can be controlled on the basis of *N*-functionalization. This enabled us to efficiently differentiate O3 and O4 to synthesize partially protected galactosamine and allosamine derivatives by epimerizing C4 and C3, respectively. We initiated the synthesis of galactosamine precursors **15** and **16** following the protocols described in Scheme 2 and acetylation using pyridine and acetic anhydride, which produced 60% and 62% of the C4 alcohols **15** and **16**, respectively. Next, we performed the regioselective protection of 7 followed by epimerization²³ using Tf₂O and NaNO₂ in DMF, which afforded 72% and 69% of **17** and **18**, respectively (Scheme 4).

The 1,6-anhydro bridges of 17 and 18 were opened by treating with $Sc(OTf)_3/Ac_2O_1^{24}$ which afforded 19 and 20 with excellent yield (Scheme 4). Through this method, we successfully synthesized D-galactosamine derivatives 19 and 20 from free glucosamine hydrochloride (1) in only 6 steps.

Scheme 4. Synthesis of Galactosamine Derivatives 19 and 20



Similarly, we developed a four-step procedure for the synthesis of allosamine derivative 23 (Scheme 5). Using the

Scheme 5. Synthesis of Allosamine Derivative 23



aforementioned protocol, **5** was regioselectively protected with 2-NAP group. The sterically crowded hydroxyl group of **21** was converted to a triflate using triflic anhydride and pyridine. However, its epimerization using NaNO₂ in DMF provided only 36% yield of **22**, whereas 17% of the triflate derivative and 18% of **21** were recovered. This indicated the reactivity of 3-OH toward epimerization is lower probably because of the steric hindrance. Furthermore, addition of extra NaNO₂ and prolonged reaction time did not have any influence on **22**. The ring was opened under acidic condition to afford acetate derivative **23** in good yield, surprisingly with free 6-OH.

Alternatively, we also developed a convenient method for protecting the C4 hydroxyl group when the C2 amine contains an N-H (Scheme 6) that was applied for the synthesis of

Scheme 6. Synthesis of Allosamine Derivative 27



allosamine derivative **27**. After the microwave-assisted 1,6cyclization of 3, the intermediate **6** was treated with Amberlie H^+ to afford **24** in 70% over 2 steps. It was subjected to regioselective benzoylation through a tin-mediated reaction. This reaction could be completed only in the presence of 0.5 equiv of Me₂SnCl₂, and afforded **25** in an excellent 95% yield after overnight stirring at room temperature. The triflation of **25** using Tf₂O and pyridine followed by the epimerization using CsOAc in DMF²⁵ resulted in 75% of **26**. The 1,6-anhydro-ring opening using Ac₂O and Sc(OTf)₃ yielded 92% of the acetate derivative **27**.

In short, our methodology using microwave-assisted heating allows the synthesis of 1,6-anhydroglucosamine derivative by using TMSOTf as a catalyst and obviates the traditional anomeric protections, which generally require 2 or 3 steps and need to be removed in the later stage of oligosaccharide synthesis. This method is straightforward and does not entail lengthy steps, and the compounds prepared from inexpensive glucosamine hydrochloride (1) are easy to handle and accessible within a short time. This transformation also renders the differentiation among all the hydroxyl groups of 2-deoxyaminosugars easier, of which the differentiation of the remaining C3 and C4 secondary hydroxyl groups can be

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controlled by the presence/absence of hydrogen bonding between the O4 and the C2 amine functionality. The synthesis of a galactosamine derivative, including regioselective protection of 1,6 anhydroglucosamine, C4-epimerization followed by ring opening under acidic condition, could be achieved in 6 steps from free glucosamine hydrochloride (1). A rare allosamine derivative could be synthesized via either the C4 regioselective reductive etherification or the regioselective tinmediated benzoylation of the 1,6-anhydroglucosamine in good overall yields.

EXPERIMENTAL SECTION

General Information. All the reactions were conducted in flamedried glassware, under nitrogen atmosphere. Methanol, acetonitrile and dichloromethane were purified and dried by using a safe purification system filled with anhydrous Al2O3. All other reagents were obtained from commercial sources and used without further purification unless otherwise mentioned. Water was distilled. Flash column chromatography was carried out on Silica Gel 60 (230-400 mesh). TLC was performed on recoated glass plates of Silica Gel 60 F254 (0.25 mm); detection was executed by spraying a solution of Ce(NH₄)₂(NO₃)₆, (NH₄)₆Mo₇O₂₄, and H₂SO₄ in water and subsequent heating on a hot plate. Specific rotations were taken at ambient temperature conditions and reported in $10^{-1} \cdot \text{deg} \cdot \text{cm}^2 \cdot \text{g}^{-1}$; the sample concentrations are in $g \cdot dL^{-1}$. The microwave-assisted reactions were performed using the Discover SP system (CEM) in the sealed reaction vessels in dynamic mode with the temperature monitored using a vertically focused IR sensor. ¹H and ¹³C NMR spectra were recorded on 400 and 600 MHz instruments. Chemical shifts are in ppm from Me₄Si, calibrated at δ 7.24 or δ 0.00 for ¹H spectra (residual CHCl₃ or TMS respectively), and δ 77.23 for ¹³C spectra. Splitting patterns were designated as follows: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad. Proton peak were assigned based on 2D NMR spectra (1H-1H COSY, HSQC, and NOESY). The Infrared Spectra were recorded using thin film method in the region 4000-600 cm⁻¹. The X-ray intensity data were measured at low temperature 100 K using Mo K α radiation diffractometer equipped with a kappa geometry goniometer and corrected for absorption effects using the Multi-Scan method (SADABS).

General Procedure for the Microwave Assisted Synthesis of 1,6-Anhydrosugars (5–7). In a dried 35 mL microwave vial, the per-O-trimethylsilylated glucosamine derivative (2.025 mmol, 1.0 equiv) was dissolved in MeCN (20 mL), followed by the addition of TMSOTf (37 μ L, 0.203 mmol, 0.1 equiv). The mixture was subjected to microwave irradiation at 100 °C for 5 min. The consumption of the starting material was confirmed by TLC. The reaction was treated with HMDS (844 μ L, 4.050 mmol, 2.0 equiv) and further stirred for 30 min at rt. The mixture was evaporated and the crude mixture was purified by short column chromatography (10% ethyl acetate/hexane) to afford the desired compound.

i,6-Anhydro-2-azido-2-deoxy-3,4-di-O-trimethylsilyl-β-D-glucopyranose (**5**).¹⁶ Colorless oil (570 mg, 85%), $[\alpha]^{28}{}_{\rm D}$ + 36.8 (*c* 1.0, CHCl₃); IR (CHCl₃) ν 2957, 2100, 1251, 1107, 1010, 839, 748 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.44 (s, 1H, H-1), 4.38 (d, *J* = 4.6 Hz, 1H, H-5), 4.09 (dd, *J* = 7.0, 1.0 Hz, 1H, H-6a), 3.69–3.71 (m, 1H, H-3), 3.67 (dd, *J* = 6.9, 5.9 Hz, 1H, H-6b), 3.49 (bt, 1H, *J* = 2.0 Hz, H-4), 2.93 (s, 1H, H-2), 0.16 (s, 9H), 0.13 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 101.1 (CH), 77.2 (CH), 73.3 (CH), 72.6 (CH), 65.3 (CH₂), 62.8 (CH), 0.2 (CH₃), 0.1 (CH₃); HRMS (ESI-TOF) *m*/*z* [M + Na]⁺ calcd for C₁₂H₂₅N₃O₄Si₂Na 354.1281, found 354.1273.

1,6-Anhydro-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)-3,4-di-O-trimethylsilyl-β-D-glucopyranose (6).¹⁶ Colorless oil (523 mg, 70%), $[\alpha]^{29}_{D}$ –31.7 (*c* 1.0, CHCl₃); IR (CHCl₃) ν 2956, 1742, 1508, 1252, 1102, 841, 734 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.64 (d, *J* = 11.0 Hz, 1H, NH), 5.35 (s, 1H, H-1), 4.70 (ABq, *J* = 12.0 Hz, 2H, OCH₂), 4.32 (d, *J* = 5.5 Hz, 1H, H-5), 4.27 (d, *J* = 7.0 Hz, 1H, H-6a), 3.68 (t, *J* = 6.4 Hz, 1H, H-6b), 3.63 (d, *J* = 11.0 Hz, 1H, H-2), 3.55 (s, 1H, H-3), 3.50 (s, 1H, H-4), 0.14 (s, 9H), 0.13 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 154.1 (C), 101.1 (CH), 100.8 (CH), 95.8 (CH), 76.7 (CH), 74.7 (CH₂), 73.3 (CH), 72.0 (CH), 65.1 (CH₂), 53.2 (CH), 0.1 (CH₃), 0.0 (CH₃); HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₁₅H₂₈Cl₃NO₆Si₂Na 502.0418, found 502.0414.

1,6-Anhydro-2-deoxy-2-trichloroacetamido-3,4-di-O-trimethylsilyl-β-D-glucopyranose (7).¹⁶ White solid (514 mg, 70%), $[a]^{29}_{\rm D}$ –33.3 (c 1.0, CHCl₃); mp 72–73 °C; IR (CHCl₃) ν 2956, 1716, 1508, 1252, 1098, 1013, 889, 839, 747 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.43 (d, *J* = 9.6 Hz, 1H, NH), 5.37 (s, 1H, H-1), 4.35 (bt, *J* = 2.7 Hz, 1H, H-5), 4.32 (d, *J* = 7.0 Hz, 1H, H-6b), 3.86 (dd, *J* = 9.6, 1.2 Hz, 1H, H-2), 3.72 (t, *J* = 6.1 Hz, 1H, H-6a), 3.55 (t, *J* = 1.6 Hz, 1H, H-3), 3.54 (d, *J* = 1.2 Hz, 1H, H-4), 0.15 (s, 9H), 0.14 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 161.4 (C), 100.2 (CH), 72.7 (CH), 71.8 (CH), 65.1 (CH₂), 52.8 (CH), 0.1 (CH₃), 0.0 (CH₃); HRMS (ESI-TOF) *m*/*z* [M + Na]⁺ calcd for C₁₄H₂₆Cl₃NO₅Si₂Na 472.0313, found 472.0322.

General Procedure for the Regioselective Protection of 1,6-Anhydrosugars and Acetylation (8-13). To a solution of compound 7 (100 mg, 0.222 mmol, 1.0 equiv) in dry CH_2Cl_2 (1.5 mL) was added 3 Å molecular sieves (0.2 g), triethylsilane (42 μ L, 0.266 mmol, 1.2 equiv) and aldehyde (0.266 mmol, 1.2 equiv) under N₂ atmosphere. The reaction mixture was stirred for 30 min at room temperature. The mixture was cooled to 0 °C. TMSOTf (4 μ L, 0.022 mmol, 0.1 equiv) was added at 0 °C and the mixture was stirred for another 30 min. Reaction was monitored by using TLC. A second portion of aldehyde (0.089 mmol, 0.4 equiv), triethylsilane (14 μ L, 0.089 mmol, 0.4 equiv) and TMSOTf (4 µL, 0.022 mmol, 0.1 equiv) was added to the reacion mixture, which was further stirred for 5 h at 0 °C. The reaction was monitored by using TLC. Upon completion, the reaction was quenched with TBAF (444 μ L, 0.444 mmol, 2.0 equiv) and stirred at room temperature for another 2 h. Pyridine (89 μ L, 1.110 mmol, 5.0 equiv) and acetic anhydride (27 μ L, 0.289 mmol, 1.3 equiv) were added sequentially, and the mixture was stirred for additional 2 h at room temperature. The reaction was monitored by TLC. Upon completion, the mixture was filtered through a Celite pad. The filtrate was evaporated and the crude was purified by column chromatography (30% ethyl acetate/hexane) to afford desired 1,6anhydrosugar derivatives (8-13).

4-O-Acetyl-1,6-anhydro-2-deoxy-3-O-(2-naphthylmethyl)-2-trichloroacetamido-β-D-glucopyranose (**8**). White solid (68 mg, 63%), $[\alpha]^{29}_{D}$ -39.7 (*c* 1.0, CHCl₃); mp 121–122 °C; IR (CHCl₃) ν 3411, 2960, 2902, 1714, 1508, 1285, 1173, 1020, 789, 704 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.78–7.83 (m, 4H, ArH), 7.44–7.47 (m, 3H, ArH), 7.12 (d, *J* = 9.5 Hz, 1H, NH), 5.50 (s, 1H, H-1), 4.86 (ABq, *J* = 12.0 Hz, 2H, ArCH₂), 4.85 (s, 1H, H-4), 4.63 (d, *J* = 5.4 Hz, 1H, H-5), 4.37 (d, *J* = 7.6 Hz, 1H, H-6b), 4.21 (d, *J* = 9.4 Hz, 1H, H-3), 3.83 (t, *J* = 6.2 Hz, 1H, H-6a), 3.48 (s, 1H, H-2), 2.07 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 169.4 (C), 161.4 (C), 134.6 (C), 133.3 (C), 133.2 (C), 128.5 (CH), 128.1 (CH), 127.8 (CH), 126.7 (CH), 126.3 (CH), 126.2 (CH), 125.6 (CH), 100.1 (CH), 92.4 (CH), 76.2 (CH), 73.8 (CH), 72.5 (CH₂), 70.4 (CH), 65.6 (CH₂), 49.6 (CH), 21.1 (CH₃); HRMS (ESI-TOF) *m*/*z* [M - H]⁻ calcd for C₂₁H₁₉Cl₃NO₆ 486.0278, found 486.0270.

4-O-Acetyl-1,6-anhydro-2-deoxy-3-O-p-methoxybenzyl-2-trichloroacetamido-β-D-glucopyranose (**10**). Colorless oil (69 mg, 66%), $[\alpha]^{28}_{D}$ –16.4 (c 0.3, CHCl₃); IR (CHCl₃) ν 3412, 2957, 2905, 1715, 1512, 1228, 1035, 889, 789, 704 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.25 (dd, *J* = 11.6, 2.6 Hz, 2H, ArH), 7.10 (d, *J* = 9.4 Hz, 1H, NH), 6.86 (dd, *J* = 9.9, 3.0 Hz, 2H, ArH), 5.46 (s, 1H, H-1), 4.78 (bs, 1H, H-4), 4.62 (ABq, *J* = 11.5 Hz, 2H, ArCH₂), 4.6 (d, *J* = 5.6 Hz, 1H, H-5), 4.32 (d, *J* = 7.4 Hz, 1H, H-6a), 4.13 (d, *J* = 9.4 Hz, 1H, H-2), 3.80 (dd, *J* = 7.1, 6.6 Hz, 1H, H-6b), 3.78 (s, 3H, OCH₃), 3.40 (bs, 1H, H-3), 2.08 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 169.4 (C), 161.2 (C), 159.6 (C), 129.4 (CH), 129.3 (CH), 114.0 (CH), 100.2 (CH), 76.0 (CH), 73.8 (CH), 72.1 (CH), 70.5 (CH), 65.5 (CH₂), 55.4 (CH), 49.7 (CH), 21.1 (CH₃); HRMS (ESI-TOF) *m*/*z* [M - H]⁻ calcd for C₁₈H₁₉Cl₃NO₇ 466.0227, found 466.0225.

4-O-Acetyl-1,6-anhydro-3-O-benzyl-2-deoxy-2-trichloroacetamido-β-D-glucopyranose (**11**). White solid (58 mg, 60%), $[\alpha]^{28}_{D}$ -50.3 (*c* 1.0, CHCl₃); mp 153–154 °C; IR (CHCl₃) ν 1714, 1507, 1225, 1019, 818, 741, 698 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.32 (d, *J* = 4.3 Hz, 4H, ArH), 7.29–7.25 (m, 1H, ArH), 7.11 (d, *J* = 9.4 Hz, NH), 5.47 (s, 1H, H-1), 4.80 (s, 1H, H-4), 4.69 (ABq, J = 11.8 Hz, 2H, ArCH₂), 4.60 (s, 1H, H-5), 4.33 (d, J = 7.4 Hz, 1H, H-6a), 4.13 (d, J = 9.4 Hz, 1H, H-2), 3.80 (dd, J = 7.0, 6.0 Hz, 1H, H-6b), 3.42 (s, 1H, H-3), 2.07 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 169.4 (C), 161.3 (C), 128.6 (CH), 128.0 (CH), 127.6 (CH), 127.1 (CH), 100.1 (CH), 76.3 (CH), 73.8 (CH), 72.4 (CH), 70.3 (CH), 65.5 (CH₂), 49.7 (CH), 21.1 (CH₃); HRMS (ESI-TOF) m/z [M – H][–] calcd for C₁₇H₁₇Cl₃NO₆ 436.0121, found 436.0125.

4-O-Acetyl-1,6-anhydro-2-deoxy-3-O-p-bromobenzyl-2-trichloroacetamido- β -D-glucopyranose (12). Colorless oil (71 mg, 62%), $[\alpha]^{29}_{D}$ -44.8 (c 1.0, CHCl₃); IR (CHCl₃) ν 3411, 2960, 2902, 1714, 1507, 1370, 1189, 1012, 924, 818, 706 cm⁻¹; ¹H NMR (400 MHz, $CDCl_3$) δ 7.46 (t, J = 2.4 Hz, 1H, ArH), 7.44 (t, J = 2.0 Hz, 1H, ArH), 7.22 (t, J = 2.3 Hz, 1H, ArH), 7.19 (t, J = 2.0 Hz, 1H, ArH), 7.26 (dd, J = 11.7, 2.6 Hz, 2H, ArH), 7.10 (d, J = 9.5 Hz, 1H, NH), 5.48 (bs, 1H, H-1), 4.80 (m, 1H, H-4), 4.65 (ABq, J = 12.8 Hz, 2H, ArCH₂), 4.62, (m, 1H, H-5), 4.30 (dd, I = 7.5, 0.85 Hz, 1H, H-6a), 4.11 (dd, I = 9.4)1.2 Hz, 1H, H-2), 3.81 (dd, J = 7.5, 5.8 Hz, 1H, H-6b), 3.40 (ddd, J = 4.4, 3.0, 1.6 Hz, 1H, H-3), 2.09 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 169.4 (C), 161.5 (C), 136.3 (C), 131.7 (CH), 129.3 (CH), 121.9 (CH), 100.0 (CH), 92.4 (CH), 76.4 (CH), 73.8 (CH), 71.6 (CH₂), 70.3 (CH), 65.5 (CH₂), 49.5 (CH), 21.1 (CH₃); HRMS (ESI-TOF) $m/z [M - H]^-$ calcd for $C_{17}H_{16}BrCl_3NO_6$ 513.9227, found 513.9235.

4-O-Acetyl-1,6-anhydro-2-deoxy-3-O-p-chlorobenzyl-2-trichloroacetamido-β-D-glucopyranose (13). Colorless oil (61 mg, 58%), [α]²⁹_D -42.3 (c 1.0, CHCl₃); IR (CHCl₃) ν 3415, 2963, 2903, 1714, 1712, 1506, 1225, 1088, 1016, 817, 705 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.31-7.24 (m, 4H, ArH), 7.11 (d, *J* = 9.3 Hz, 1H, NH), 5.48 (s, 1H, H-1), 4.80 (d, *J* = 1.3 Hz, 1H, H-4), 4.66 (ABq, *J* = 11.9 Hz, 2H, ArCH₂), 4.62 (m, 1H, H-5), 4.30 (dd, *J* = 7.5, 0.8 Hz, 1H, H-6a), 4.11 (dd, *J* = 9.5, 1.3 Hz, 1H, H-2), 3.82 (dd, *J* = 7.4, 5.8 Hz, 1H, H-6b), 3.40 (ddd, *J* = 4.4, 3.0, 1.5 Hz, 1H, H-3), 2.09 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 169.5 (C), 161.5 (C), 135.8 (C), 133.8 (C), 129.0 (CH), 128.8 (CH), 100.0 (CH), 92.4 (CH), 76.3 (CH), 73.8 (CH), 71.6 (CH₂), 70.3 (CH), 65.5 (CH₂), 49.5 (CH), 21.1 (CH₃); HRMS (ESI-TOF) *m*/*z* [M - H]⁻ calcd for C₁₇H₁₆Cl₄NO₆ 469.9732, found 469.9724.

3-O-Acetyl-1,6-anhydro-2-azido-2-deoxy-4-O-(2-naphthylmethyl)- β -D-qlucopyranose (14). To a solution of 5 (100 mg, 0.302 mmol, 1.0 equiv) in dry CH₂Cl₂ (1.5 mL) was added 3 Å molecular sieves (0.2 g), triethylsilane (58 µL, 0.362 mmol, 1.2 equiv) and 2naphthaldehyde (56 mg, 0.362 mmol, 1.2 equiv) under N₂ atmosphere. The mixture was stirred for 30 min at room temperature. The mixture was cooled to -78 °C. TMSOTf (5 μ L, 0.030 mmol, 0.1 equiv) was added at -78 °C and the mixture was stirred for another 30 min. Reaction was monitored by using TLC. A second portion of 2naphthaldehyde (19 mg, 0.121 mmol, 0.4 equiv), triethylsilane (19 µL, 0.121 mmol, 0.4 equiv) and TMSOTf (5 μ L, 0.030 mmol, 0.1 equiv) was added to the reacion mixture, which was further stirred for 16 h at -78 °C. The reaction was monitored by using TLC. Upon completion, the reaction was quenched with TBAF (604 μ L, 0.604 mmol, 2.0 equiv) and stirred at room temperature for another 2 h. Pyridine (122 μ L, 1.510 mmol, 5.0 equiv) and acetic anhydride (37 μ L, 0.393 mmol, 1.3 equiv) were added sequentially, and the mixture was stirred for additional 2 h at room temperature. The reaction was monitored by TLC. Upon completion, the mixture was filtered through a Celite pad. The filtrate was evaporated and the crude was purified by column chromatography (30% ethyl acetate/hexane) to afford desired compound 14. White solid (80 mg, 72%), $[\alpha]^{29}_{D}$ + 66.2 (c 1.0, CHCl₃); mp 92–93 °C; IR (CHCl₃) v 2962, 2903, 2153, 2099, 1736, 1219, 1023, 817, 748 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.84-7.80 (m, 4H, ArH), 7.52 (dd, J = 8.4, 1.5 Hz, 1H, ArH), 7.49-7.44 (m, 2H, ArH), 5.50 (s, 1H, H-1), 5.09 (t, J = 1.4 Hz, 1H, H-3), 4.91 (ABq, J = 12.7 Hz, 2H, ArCH₂), 4.60 (d, J = 5.5 Hz, 1H, H-5), 3.89 (dd, J = 7.6, 0.9 Hz, 1H, H-6a), 3.71 (dd, J = 7.4, 5.8 Hz, 1H, H-6b), 3.29 (d, J = 0.76 Hz, 1H, H-4), 3.20 (s, 1H, H-2), 2.06 (s, 3H, CH₃); ^{13}C NMR (100 MHz, CDCl₃) δ 169.8 (C), 134.8 (C), 133.3 (C), 133.2 (C), 128.6 (CH), 128.0 (CH), 127.8 (CH), 127.2 (CH), 126.4 (CH), 126.2 (CH), 126.0 (CH), 100.2 (CH), 74.7 (CH), 73.78

(CH), 73.77 (CH), 71.5 (CH₂), 69.6 (CH), 65.2 (CH₂), 59.1 (CH), 21.2 (CH₃); HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₁₉H₁₉N₃O₅Na 392.1222, found 392.1223.

General Procedure for the Regioselective Protection of 1,6-Anhydrosugars (15 and 16). To a solution of compound 7 (100 mg, 0.222 mmol, 1.0 equiv) in dry CH₂Cl₂ (1.5 mL) was added 3 Å molecular sieves (200 mg), triethylsilane (42 µL, 0.266 mmol, 1.2 equiv) and aldehyde (0.266 mmol, 1.2 equiv) under N₂ atmosphere. The reaction mixture was stirred for 30 min at room temperature. The mixture was cooled to 0 °C. TMSOTf (4 μ L, 0.022 mmol, 0.1 equiv) was added at 0 °C and the mixture was stirred for another 30 min. Reaction was monitored by using TLC. A second portion of aldehyde (0.36 equiv), triethylsilane (14 μ L, 0.089 mmol, 0.4 equiv) and TMSOTf (4 μ L, 0.022 mmol, 0.1 equiv) was added to the reacion mixture, which was further stirred for 5 h at 0 °C. The reaction was monitored by using TLC. Upon completion, the reaction was quenched with TBAF (444 µL, 0.444 mmol, 2.0 equiv) and stirred at room temperature for another 2 h. The reaction mixture was then filtered by passing through a short pad Celite bed. The filtrate was concentrated and purified by column chromatography (30% ethyl acetate/hexane) to afford the desired product.

1,6-Anhydro-3-O-benzyl-2-deoxy-2-trichloroacetamido-β-D-glucopyranose (**15**). Colorless oil (53 mg, 60%), ¹H NMR (400 MHz, CDCl₃) δ 7.36–7.26 (m, 6H, ArH, NH), 5.45 (s, 1H, H-1), 4.65 (ABq, *J* = 12.0 Hz, 2H, ArCH₂), 4.51 (s, 1H, H-5), 4.30 (d, *J* = 7.2 Hz, 1H, H-6a), 4.13 (d, *J* = 9.4 Hz, 1H, H-2), 3.8 (s, 1H, H-4), 3.76 (d, *J* = 6.6 Hz, 1H, H-6b), 3.45 (s, 1H, H-3), 2.38 (bs, 1H, OH); ¹³C NMR (100 MHz, CDCl₃) δ 161.6 (C), 137.5 (C), 128.6 (CH), 128.0 (CH), 127.8 (CH), 100.2 (CH), 92.4 (CH), 78.2 (CH), 76.2 (CH), 72.2 (CH₂), 69.6 (CH), 65.2 (CH₂), 49.4 (CH); HRMS (ESI-TOF) *m*/*z* [M + Na]⁺ calcd for C₁₅H₁₆Cl₃NO₅Na 417.9992 found 417.9999.

1,6-Anhydro-2-deoxy-3-O-(2-naphthylmethyl)-2-trichloroacetamido-β-D-glucopyranose (**16**). White solid (61 mg, 62%), $[\alpha]^{27}_{\rm D}$ –96.6 (*c* 0.4, CHCl₃); mp 116–117 °C; IR (CHCl₃) ν 1373, 1317, 1094, 1013, 924, 819, 784, 754 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ 7.82–7.81 (m, 3H, ArH), 7.77 (s, 1H, ArH), 7.48–7.44 (m, 3H, ArH), 7.30 (d, *J* = 9.0 Hz, 1H, NH), 5.48 (s, 1H, H-1), 4.82 (ABq, *J* = 12.0 Hz, 2H, ArCH₂), 4.53 (d, *J* = 5.4 Hz, 1H, H-5), 4.34 (d, *J* = 7.6 Hz, 1H, H-6a), 4.20 (d, *J* = 9.4 Hz, 1H, H-2), 3.80 (q, *J* = 5.8 Hz, 2H, H-4, H-6b), 3.49 (s, 1H, H-3), 2.01 (d, *J* = 4.2 Hz, 1H, OH); ¹³C NMR (150 MHz, CDCl₃) δ 161.7 (C), 134.9 (C), 133.4 (C), 133.3 (C), 128.6 (CH), 128.2 (CH), 127.9 (CH), 127.0 (CH), 126.4 (CH), 126.3 (CH), 125.9 (CH), 100.3 (CH), 92.5 (CH), 78.1 (CH), 76.3 (CH), 72.3 (CH₂), 69.8 (CH), 65.3 (CH₂), 49.4 (CH); HRMS (ESI-TOF) *m*/*z* [M – H]⁻ calcd for C₁₉H₁₇Cl₃NO₅ 444.0172, found 444.0178.

1,6-Anhydro-3-O-benzyl-2-deoxy-2-trichloroacetamido- β -D-galactopyranose (17). Trifluoromethanesulfonic anhydride (0.12 mL, 0.726 mmol, 2.4 equiv) was added to a solution of compound 15 (120 mg, 0.302 mmol, 1.0 equiv) in pyridine (0.10 mL, 1.268 mmol, 4.2 equiv) and dry DCM (2 mL) at 0 °C. After stirring for 30 min, the reaction was monitored by TLC. Upon completion, DCM was evaporated and the crude material was used directly in the next reaction. To a solution of the crude triflate derivative in DMF (2 mL) was added sodium nitrite (0.21 g, 3.020 mmol, 10.0 equiv) at room temperature and the mixture was stirred overnight at same temperature. The reaction was diluted with water and extracted with ethyl acetate (15 mL \times 3). The combined organic layers were dried over anhydrous MgSO4, filtered, and concentrate under a reduced pressure. The residue was purified by column chromatography (30% ethyl acetate/hexane) to give 17 as a colorless oil. Colorless oil (86 mg, 72%, over two steps), IR (CHCl₃) v 2957, 2100, 1251, 1107, 1010, 839, 748 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.40 (d, J = 7.3 Hz, 2H, ArH), 7.33 (t, J = 7.0 Hz, 2H, ArH), 7.27 (t, J = 7.3 Hz, 1H, ArH), 6.83 (d, J = 8.4 Hz, 1H, NH), 5.46 (s, 1H, H-1), 4.76 (ABq, J = 12.0 Hz, 2H, ArCH₂), 4.61 (d, J = 7.4 Hz, 1H, H-6a), 4.50 (t, J = 4.2Hz, 1H, H-5), 4.27 (d, J = 8.4 Hz, 1H, H-2), 4.21 (t, J = 4.1 Hz, 1H, H-4), 3.75 (t, J = 6.4 Hz, 1H, H-6b), 3.68 (dd, J = 3.3, 0.97 Hz, 1H, H-3); ¹³C NMR (100 MHz, CDCl₃) δ 161.5 (C), 137.4 (C), 128.2 (CH), 127.6 (CH), 99.8 (CH), 91.9 (CH), 75.9 (CH), 75.6 (CH),

73.3 (CH₂), 64.8 (CH₂), 54.8 (CH), 53.1 (CH); HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₁₅H₁₆Cl₃NO₅Na 417.9992, found 417.9988.

1,6-Anhydro-2-deoxy-3-O-(2-naphthylmethyl)-2-trichloroacetamido- β -D-galactopyranose (18). Trifluoromethan-esulfonic anhydride (0.17 mL, 0.983 mmol, 2.2 equiv) was added to a solution of compound 16 (200 mg, 0.447 mmol, 1.0 equiv) in pyridine (0.15 mL, 1.877 mmol, 4.2 equiv) and dry DCM (2 mL) at 0 °C. After stirring for 30 min, the reaction was monitored by TLC. Upon completion, the DCM was evaporated and the crude material was used without further purification for the next reaction. To a solution of the triflate derivative in DMF (2 mL) was added sodium nitrite (0.30 g, 4.47 mmol, 10.0 equiv) at room temperature, and the mixture was stirred overnight at the same temperature. The reaction was quenched with water (15 mL) and the mixture was extracted with ethyl acetate (15 mL \times 3). The layers were combined, dried over anhydrous MgSO₄, filtered, and evaporated under a reduced pressure. The residue was purified by column chromatography (30% ethyl acetate/hexane) to give compound 18. White solid (138 mg, 69% over 2 steps), $[\alpha]_{D}^{27}$ -100.5 (c 0.5, CHCl₃); mp 131-132 °C; IR (CHCl₃) ν 1991, 1698, 1575, 1396, 1317, 1027, 937, 820, 753 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) & 7.85-7.81 (m, 4H, ArH), 7.49-7.45 (m, 3H, ArH), 6.83 (d, J = 8.5 Hz, 1H, NH), 5.44 (s, 1H, H-1), 4.91 (ABq, J = 11.7 Hz, 2H, ArCH₂), 4.43 (t, J = 4.6 Hz, 1H, H-5), 4.34 (d, J = 8.5 Hz, 1H, H-2), 4.31 (d, J = 7.6 Hz, 1H, H-6a), 3.9 (pentet, J = 9.4, 5.0 Hz, 1H, H-4), 3.74 (d, J = 5.3 Hz, 1H, H-3), 3.71 (t, J = 6.2 Hz, 1H, H-6b), 3.07 (d, J = 9.5 Hz, 1H, OH); ¹³C NMR (150 MHz, CDCl₃) δ 161.9 (C), 134.3 (C), 133.4 (C), 133.3 (C), 128.8 (CH), 128.2 (CH), 127.9 (CH), 127.6 (CH), 126.5 (CH), 126.4 (CH), 125.9 (CH), 99.7 (CH), 92.1 (CH), 75.5 (CH), 75.1 (CH), 72.8 (CH₂), 64.7 (CH), 64.3 (CH₂), 51.7 (CH); HRMS (ESI-TOF) m/z [M - H]⁻ calcd for C₁₉H₁₇Cl₃NO₅ 444.0172, found 444.0169.

1,4,6-Tri-O-acetyl-3-O-benzyl-2-deoxy-2-trichloroacetamidogalactopyranose (19). To a stirred solution of compound 17 (100 mg, 0.252 mmol, 1.0 equiv) in dry CH_2Cl_2 (1.5 mL) was added Ac_2O (238 μ L, 2.52 mmol, 10.0 equiv) at room temperature. $Sc(OTf)_3$ (12.4 mg, 0.0252 mmol, 0.1 equiv) was added to the above mixture, which was stirred 4 h at room temperature. The reaction was monitored by TLC. Upon completion, the mixture was evaporated and was directly loaded on column and the chromatography (50% ethyl acetate/ hexane) afforded the desired product in 95% yield.

1,4,6-Tri-O-acetyl-3-O-benzyl-2-deoxy-2-trichloroacetamido-β-Dgalactopyranose (**19**β). Colorless oil, $[\alpha]^{29}_{D}$ –0.13 (*c* 0.6, CHCl₃); IR (CHCl₃) ν 3361, 3032, 2919, 1742, 1520, 1315, 1215, 1140, 1045, 880, 728 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.32–7.24 (m, 5H, ArH), 6.61 (d, *J* = 8.7 Hz, 1H, NH), 5.85 (d, *J* = 8.8 Hz, 1H, H-1), 5.56 (d, *J* = 3.0 Hz, 1H, H-4), 4.70 (d, *J* = 11.5 Hz, 1H, ArCH₂), 4.38 (d, *J* = 11.5 Hz, 1H, ArCH₂), 4.20 (dd, *J* = 11.3, 6.0 Hz, 1H, H-6b), 4.12–4.03 (m, 2H, H-2, H-6a), 3.96 (t, *J* = 6.7 Hz, 1H, H-5), 3.89 (dd, *J* = 10.8, 3.3 Hz, 1H, H-3), 2.14 (s, 3H, CH₃), 2.07 (s, 3H, CH₃), 2.05 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 170.4 (C), 170.1 (C), 169.1 (C), 162.0 (C), 136.7 (C), 128.6 (CH × 2), 128.4 (CH × 2), 128.3 (CH), 91.8 (CH), 74.7 (CH), 72.2 (CH), 71.6 (CH₂), 64.9 (CH), 61.8 (CH₂), 53.4 (CH), 20.7 (CH₃); HRMS (ESI-TOF) *m*/*z* [M + Na]⁺ calcd for C₂₁H₂₄Cl₃NO₉Na 562.0414, found 562.0416.

1,4,6-Tri-O-acetyl-3-O-benzyl-2-deoxy-2-trichloroacetamido-αgalactopyranose (**19**α). Colorless oil, $[α]^{29}_{D}$ + 92.6 (*c* 1.0, CHCl₃); IR (CHCl₃) ν 3361, 3032, 2919, 1742, 1520, 1315, 1215, 1140, 1045, 880, 728 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.35–7.28 (m, 5H, ArH), 6.36 (d, *J* = 3.6 Hz, 1H, H-1), 6.32 (d, *J* = 7.4 Hz, 1H, NH), 5.62 (d, *J* = 2.8 Hz, 1H, H-4), 4.57 (ABq, *J* = 11.9 Hz, 2H, ArCH₂), 4.37–4.42 (m, 1H, H-2), 4.05–4.19 (m, 3H, H-5, H-6a, H-6b), 3.77 (dd, *J* = 11.0, 3.04 Hz, 1H, H-3), 2.15 (s, 3H, CH₃), 2.05 (s, 3H, CH₃), 2.04 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 170.4 (C), 170.2 (C), 168.4 (C), 161.8 (C), 136.8 (C), 128.7 (CH), 128.2 (CH), 128.1 (CH), 92.1 (CH), 90.2 (CH), 72.4 (CH), 70.9 (CH₂), 68.8 (CH), 64.8 (CH), 61.7 (CH₂), 50.1 (CH), 20.6 (CH₃), 20.6 (CH₃); HRMS (ESI-TOF) *m*/*z* [M + Na]⁺ calcd for C₂₁H₂₄Cl₃NO₉Na 562.0414, found 562.0416.

1,4,6-Tri-O-acetyl-2-deoxy-3-O-(2-naphthylmethyl)-2-trichloroacetamido- α -p-galactopyranose (**20**). The general procedure was similar as compound 19. White solid (95 mg, 72%), $[\alpha]^{27}_{D}$ -16.75 (c 0.5, CHCl₂); mp 88-89 °C; IR (CHCl₂) ν 3311, 2923, 1507, 1472, 1396, 1224, 1017, 818, 753 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ 7.79-7.82 (m, 3H, ArH), 7.74 (s, 1H, ArH), 7.46-7.51 (m, 2H, ArH), 7.41 (dd, J = 1.4, 8.4 Hz, 1H, ArH), 6.36 (d, J = 3.7 Hz, 1H, H-1), 6.33 (d, *J* = 7.6 Hz, 1H, NH), 5.67 (d, *J* = 2.8 Hz, 1H, H-4), 4.74 (ABq, *J* = 12.2 Hz, 2H, ArCH₂), 4.44 (ddd, J = 3.8, 7.6, 11.2 Hz, 1H, H-2), 4.2-4.14 (m, 2H, H-5, H-6a), 4.08 (ddd, I = 5.4, 9.5, 17.9 Hz, 1H, H-6b), 3.82 (dd, J = 3.1, 11.1 Hz, 1H, H-3), 2.18 (s, 3H, CH₃), 2.06 (s, 3H, CH₃), 2.01 (s, 3H, CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 170.7 (C), 170.4 (C), 168.6 (C), 162.0 (C), 134.7 (C), 133.3 (CH), 128.9 (CH), 128.0 (CH), 127.9 (CH), 127.3 (CH), 126.6 (CH), 126.5 (CH), 125.8 (CH), 92.3 (CH), 90.4 (CH), 72.5 (CH), 71.1 (CH₂), 69.0 (CH), 65.0 (CH), 62.0 (CH₂), 50.3 (CH), 20.9 (CH₃), 20.9 (CH₃); HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₂₅H₂₆NO₉Cl₃Na 612.0571, found 612.0578.

1,6-Anhydro-2-azido-2-deoxy-3-O-(2-naphthylmethyl)- β -D-allopyranose (21). To a solution of compound 5 (100 mg, 0.302 mmol, 1.0 equiv) in dry DCM (1.5 mL) was added 3 Å molecular sieves (0.2 g), triethylsilane (58 μ L, 0.362 mmol, 1.2 equiv) and 2naphthaldehyde (56 mg, 0.362 mmol, 1.2 equiv) under N₂ atmosphere. The reaction mixture was stirred for 30 min at room temperature. The mixture was cooled to -78 °C. TMSOTf (5 μ L, 0.030 mmol, 0.1 equiv) was added at -78 °C and the mixture was stirred for another 30 min. Reaction was monitored by using TLC. A second portion of 2-naphthaldehyde (17 mg, 0.109 mmol, 0.36 equiv), triethylsilane (17 μ L, 0.109 mmol, 0.36 equiv) and TMSOTf (5 μ L, 0.030 mmol, 0.1 equiv) was added to the reacion mixture, which was further stirred for 16 h at -78 °C. The reaction was monitored by using TLC. Upon completion, the reaction was quenched with TBAF (604 μ L, 0.604 mmol, 2.0 equiv) and stirred at room temperature for another 2 h. The reaction mixture was then filtered by passing through a short pad Celite bed. The filtrate was concentrated and purified by column chromatography (50% ethyl acetate/hexane) to afford the desired product as a colorless oil. Colorless oil (67 mg, 68%), $[\alpha]^{27}$ -61.47 (c 0.5, CHCl₃); IR (CHCl₃) ν 1942, 1920, 1828, 1771, 1418, 1258, 1091, 1020, 922, 818 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ 7.85-7.80 (m, 4H, ArH), 7.51-7.46 (m, 3H, ArH), 5.47 (s, 1H, H-1), 4.88 (ABq, J = 12.4 Hz, 2H, ArCH₂), 4.62 (d, J = 5.2 Hz, 1H, H-5), 3.94-3.91 (m, 2H, H-3, H-6a), 3.66 (dd, J = 5.4, 7.6 Hz, 1H, H-6b), 3.42 (d, J = 2.8 Hz, 1H, H-4), 3.22 (d, J = 3.1 Hz, 1H, H-2), 2.37 (d, J = 6.3 Hz, 1H, OH); ¹³C NMR (150 MHz, CDCl₃) δ 134.8 (C), 133.2 (C), 133.1 (C), 128.5 (CH), 127.8 (CH), 127.7 (CH), 126.8 (CH), 126.3 (CH), 126.1 (CH), 125.7 (CH), 101.0 (CH), 78.4 (CH), 75.1 (CH), 72.0 (CH₂), 70.5 (CH), 66.2 (CH₂), 62.7 (CH); HRMS (ESI-TOF) m/z [M - H]⁻ calcd for C₁₇H₁₆N₃O₄ 326.1141, found 326.1147.

1,6-Anhydro-2-azido-2-deoxy-4-O-(2-naphthylmethyl)- β -D-allopyranose (22). Trifluoromethanesulfonic anhydride (0.22 mL, 1.346 mmol, 2.2 equiv) was added to a solution of compound 21 (200 mg, 0.612 mmol, 1.0 equiv) in pyridine (0.21 mL, 2.570 mmol, 4.2 equiv) and dry DCM (2 mL) at 0 °C. After stirring for 30 min, the reaction was monitored by TLC. Upon completion, the DCM was evaporated and the crude material was used without further purification for the next reaction. To a solution of the triflate derivative in DMF (2 mL) was added sodium nitrite (0.42 g, 6.120 mmol, 10 equiv) at room temperature, and the mixture was stirred overnight at the same temperature. The reaction was quenched with water (15 mL) and the mixture was extracted with ethyl acetate (15 mL \times 3). The layers were combined, dried over anhydrous MgSO4, filtered, and evaporated under a reduced pressure. The residue was purified by column chromatography (30% ethyl acetate/hexane) to give 22 as colorless oil. Colorless oil (72 mg, 36% over 2 steps), $[\alpha]_{D}^{27}$ + 12.81 (c 0.5, CHCl₃); IR (CHCl₃) v 2922, 2101, 1844, 1810, 1662, 1143, 1121, 1100, 901, 753 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.84–7.80 (m, 4H, ArH), 7.52-7.46 (m, 3H, ArH), 5.53 (s, 1H, H-1), 5.21 (s, 1H, H-4), 4.92 (ABq, J = 12.6 Hz, 2H, ArCH₂), 4.61 (d, J = 5.4 Hz, 1H, H-5), 3.90 (d, J = 7.5 Hz, H-6a), 3.72 (t, J = 6.4 Hz, 1H, H-6b), 3.31 (s, 1H, H-3), 3.20 (s, 1H, H-2); ¹³C NMR (150 MHz, CDCl₃) δ 134.4 (C), 133.1 (C), 133.1 (C), 128.5 (CH), 127.9 (CH), 127.7 (CH), 127.1

(CH), 126.3 (CH), 126.1 (CH), 125.7 (CH), 100.0 (CH), 74.5 (CH), 73.5 (CH), 71.5 (CH₂), 69.1 (CH), 65.1 (CH₂), 58.8 (CH); HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₁₇H₁₇N₃O₄Na 350.1117, found 350.1111.

1,3-Di-O-acetyl-2-azido-2-deoxy-4-O-(2-naphthylmeth-yl)-β-D-allopyranose (23). The general procedure was similar as compound 19. Colorless oil (87 mg, 66%), IR (CHCl₃) v 2924, 2108, 1652, 1602, 1340, 1214, 1160, 1008, 857, 820, 751 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) & 8.09 (s, 1H, ArH), 8.04 (s, 1H, ArH), 7.83-7.80 (m, 4H, ArH), 7.70 (s, 1H, ArH), 7.67 (s, 1H, ArH), 7.50-7.45 (m, 3H, ArH), 7.37-7.32 (m, 2H, ArH), 6.26 (d, J = 3.6 Hz, 1H, H-1), 5.58 (d, J = 10.0 Hz, 1H, H-3), 5.55 (d, J = 8.6 Hz, 1H, H-3), 4.74 (ABq, J = 11.2 Hz, 2H, CH₂Ar), 4.34-4.19 (m, 4H, H-6a, H-6b, H-6a, H-6b), 3.98 (dt, J = 3.0, 10.0 Hz, 1H, H-5, H-5), 3.73 (t, J = 9.4 Hz, 1H, H-4),3.69-3.68 (m, 1H, H-4), 3.47 (dd, J = 3.6, 10.6 Hz, 1H, H-2), 2.16 (s, 3H, CH₃), 2.15 (s, 3H, CH₃), 1.92 (s, 3H, CH₃), 1.91 (s, 3H, CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 170.3 (C), 168.5 (C), 168.4 (C), 159.7 (C), 159.6 (C), 134.0 (C), 133.1 (C), 133.0 (C), 128.5 (CH), 127.9 (CH), 127.7 (CH), 127.5 (CH), 127.2 (CH), 126.4 (CH), 126.3 (CH), 125.9 (CH), 92.7 (CH), 90.2 (CH), 75.1 (CH), 74.9 (CH), 74.8 (CH), 74.7 (CH), 74.1 (CH₂), 73.8 (CH), 72.0 (CH), 70.9 (CH), 62.8 (CH), 62.0 (CH₂), 60.4 (CH), 29.6 (CH₃), 20.9 (CH_3) , 20.8 (CH_3) , 20.5 (CH_3) ; HRMS (ESI-TOF) $m/z [M + Na]^+$ calcd for C₂₁H₂₃N₃O₇Na 452.1434, found 452.1428.

1,6-Anhydro-2-deoxy-2-(2,2,2-trichloroethoxycarbo-nylamino)- β -D-glucopyranose (24). In a over dried 35 mL microwave vial per-Otrimethylsilvlated glucosamine derivative 3 (1.0 g, 2.024 mmol, 1.0 equiv) was dissolved in MeCN (20 mL), followed by the addition of 10 mol % of TMSOTf. The resulting mixture was subjected to microwave irradiation at 100 °C for 5 min. TLC confirmed the consumption of starting material. Upon completion, Amberlite 120 H⁺ resin was added and the mixture was stirred for additional 10 min at room temperature. The crude reaction mixture was purified by using a short flash column on silica gel (70% ethyl acetate/hexane) to afford desired product. White solid (366 mg, 70%), $[\alpha]^{30}_{D}$ –14.0 (c 1.0, MeOH); mp 138–139 °C; IR (CHCl₃) v 3430, 1720, 1515, 1268, 1109, 1021, 920, 712 cm⁻¹; ¹H NMR (400 MHz, MeOD) δ 5.37 (s, 1H, H-1), 4.85 (d, J = 2.4 Hz, 2H, TrocCH₂), 4.55 (d, J = 5.4 Hz, 1H, H-5), 4.25 (d, J = 7.2 Hz, 1H, H-6a), 3.74 (dd, J = 7.0, 6.0 Hz, 1H, H-6b), 3.69 (m, 1H, H-4), 3.66 (s, 1H, H-2), 3.61 (s, 1H, H-3); ¹³C NMR (100 MHz, MeOH) δ 156.0 (C), 102.4 (C), 77.8 (CH), 75.6 (CH₂), 73.6 (CH), 72.5 (CH), 66.4 (CH₂), 55.3 (CH); HRMS (ESI-TOF) m/z [M - H]⁻ calcd for C₉H₁₁Cl₃NO₆ 333.9652, found 333.9655.

1,6-Anhydro-4-O-benzoyl-2-deoxy-2-(2,2,2-trichloro-ethoxycarbonylamido)- β -D-qlucopyranose (25). To a stirred solution of diol 24 (100 mg, 0.297 mmol, 1.0 equiv) in dry THF (1.5 mL) was added triethylamine (207 μ L, 1.485 mmol, 5.0 equiv) at room temperature. A clear solution observed after 10 min. Me₂SnCl₂ (32 mg, 0.148 mmol, 0.5 equiv) was added to the above mixture, and white suspension was observed in the reaction. To the above suspension, benzoyl chloride (44.85 μ L, 0.386 mmol, 1.3 equiv) was added dropwise. The reaction mixture turned slightly pink color solution and was stirred overnight at room temperature. The reaction was monitored by TLC. The reaction was quenched with 1 N HCl and was extracted with CH₂Cl₂. The organic layers were combined, dried over anhydrous MgSO4, filtered, concentrated and chromatographed (30% ethyl acetate/hexane) to afford the desired product 25 in 95% yield. White solid (124 mg, 95%), [α]²⁹_D –95.4 (c 1.0, CHCl₃); mp 149–150 °C; IR (CHCl₃) ν 3430, 1720, 1515, 1268, 1109, 1021, 920, 712 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.02 (d, J = 8.0 Hz, 2H, ArH), 7.58 (t, J = 7.0 Hz, 1H, ArH), 7.44 (t, J = 7.3 Hz, 2H, ArH), 5.63 (d, J = 9.7 Hz, 1H, NH), 5.54 (s, 1H, H-1), 5.00 (s, 1H, H-4), 4.77 (d, J = 5.6 Hz, 1H, H-3), 4.70 (ABq, J = 11.5 Hz, 2H, Troc CH₂), 4.40 (d, J = 7.5 Hz, 1H, H-6a), 3.90 (d, J = 5.1 Hz, 1H, H-5), 3.87-3.83 (m, 2H, H-2, H-6b), 3.29 (bs, 1H, OH); 13 C NMR (100 MHz, CDCl₃) δ 165.4 (C), 154.0 (C), 133.8 (C), 129.8 (CH), 129.7 (CH), 128.8 (CH), 101.0 (CH), 95.4 (CH), 74.9 (CH₂), 74.1 (CH), 72.5 (CH), 70.6 (CH), 66.0 (CH₂), 53.0 (CH); HRMS (ESI-TOF) m/z [M - H]⁻ calcd for C₁₆H₁₅NO₇Cl₃ 437.9914, found 437.9919.

3-O-Acetyl-1,6-anhydro-4-O-benzoyl-2-deoxy-2-(2,2,2-trichloroethoxycarbonylamino)- β -D-allopyranose (26). Trifluoromethanesulfonic anhydride (85 mL, 0.5 mmol, 2.2 equiv) was added to a solution of compound 25 (100 mg, 0.227 mmol, 1.0 equiv) in pyridine (78 mL, 0.953 mmol, 4.2 equiv) and dry CH₂Cl₂ (1.5 mL) at 0 °C. After 30 min, the reaction was monitored by TLC. Upon completion, the DCM was ecaporated and the crude material used directly for the next reaction. To a solution of the triflate derivative in DMF (1 mL) was added CsOAc (871 mg, 4.54 mmol, 20.0 equiv) at room temperature and the mixture was stirred overnight at same temperature. The mixture was diluted with water (10 mL) and extracted with ethyl acetate (15 mL × 3). The organic layers were combined, dried over anhydrous MgSO₄, filtered, and evaporated under a reduced pressure. The residue was purified by column chromatography (30% ethyl acetate/hexane) to afford 26 in 75% yield.

Note: In the case of using $NaNO_2$ for the epimerization, the spots of the desired compound and starting material were very close and difficult to isolate; however the reaction with CsOAc provided good yield and purification becomes easier.

Colorless oil (82 mg, 75%), $[\alpha]^{28}_{D}$ + 3.0 (*c* 0.15, CHCl₃); IR (CHCl₃) ν 2957, 2100, 1251, 1107, 1010, 839, 748 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.05 (d, J = 8.8 Hz, 2H, ArH), 7.60 (t, J = 7.4 Hz, 1H, ArH), 7.48 (t, J = 7.7 Hz, 2H, ArH), 5.61 (d, J = 10.4 Hz, 1H, NH), 5.53 (d, J = 2.5 Hz, 1H, H-1), 5.35–5.32 (m, 2H, H-3, H-4), 4.92 (d, J = 12.2 Hz, 1H, TrocCH₂), 4.81 (dd, J = 5.0, 1.7 Hz, 1H, H-5), 4.55 (d, J = 11.8 Hz, 1H, TrocCH₂), 4.28–4.24 (m, 1H, H-2), 4.03 (d, J = 8.4 Hz, 1H, H-6b), 3.86 (dd, J = 8.3, 5.4 Hz, 1H, H-6a), 1.93 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 169.6 (C), 165.7 (C), 154.5 (C), 133.8 (C), 129.8 (CH), 129.3 (CH), 129.0 (CH), 101.1 (CH), 74.8 (CH₂), 74.3 (CH), 70.4 (CH), 65.8 (CH₂), 63.5 (CH), 52.4 (CH), 20.7 (CH₃); HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₁₈H₁₈Cl₃NO₈Na 503.9996, found 503.9985.

1,3,6-Tri-O-acetyl-4-O-benzoyl-2-deoxy-2-(2,2,2-trichlo-roethoxycarbonylamido)- β -D-allopyranose (27). To a stirred solution of compound 26 (100 mg, 0.226 mmol, 1.0 equiv) in dry CH_2Cl_2 (1.5 mL) was added Ac₂O (21 μ L, 2.26 mmol, 10.0 equiv) at room temperature. Sc(OTf)₃ (11 mg, 0.0226 mmol, 0.1 equiv) was added to the above mixture, which was stirred overnight at room temperature. The reaction was monitored by TLC. Upon completion, the mixture was evaporated and was directly loaded on column and the chromatography (30% ethyl acetate/hexane) afforded the desired product in 92% yield. Colorless oil (112 mg, 92%), IR (CHCl₃) v 3319, 2959, 1728, 1214, 1177, 1042, 1026, 821, 759, 709 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.89 (dd, J = 7.6 Hz, 3H, ArH), (t, J = 7.5 Hz, 1H, ArH), 7.41 (t, J = 7.8 Hz, 3H, ArH), 6.20 (d, J = 4.1 Hz, 1H, H-1 α), 5.96 (d, J = 8.1 Hz, 1H, H-1 β), 5.80–5.78 (m, 2H, H-3 α , H- 3β), 5.27–5.19 (m, 3H, H-4a, H-4b, H6a α), 4.83 (d, J = 12.0 Hz, 1H, Troc CH₂), 4.71 (d, J = 9.8 Hz, 1H, Troc CH₂), 4.60 (d, J = 12.0 Hz, 1H, Troc CH₂), 4.43–4.38 (m, 1H, H-2 α), 4.33–4.18 (m, 6H, H-2 β , H-5β, H-6aβ, H-6bβ, H-5α, H-6α), 2.18 (s, 3H, CH₃), 2.12 (s, 3H, CH₃), 2.12 (s, 3H, CH₃ × 2), 2.03 (s, 3H, CH₃), 2.01 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ 170.5 (C), 169.7 (C), 169.5 (C), 169.3 (C), 168.9 (C), 164.7 (C), 164.5 (C), 153.8 (C), 153.6 (C). 133.6 (C), 129.8 (CH), 129.5 (CH), 128.9 (CH), 128.6 (CH), 95.3 (CH), 91.5 (CH), 89.4 (CH), 75.0 (CH2), 74.8 (CH2), 71.6 (CH), 69.0 (CH), 68.0 (CH), 67.1 (CH), 66.5 (CH), 62.4 (CH₂), 62.1 (CH₂), 51.9 (CH), 49.2 (CH), 21.0 (CH₃), 20.9 (CH₃ × 2), 20.7 (CH₃), 20.6 (CH₃); HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₂₂H₂₄NO₁₁NaCl₃ 606.0313, found 606.0305.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b02038.

¹H and ¹³C NMR and HRMS spectra of all new compounds (PDF)

Crystallographic data for 7 (CCDC 1498510) (CIF) Crystallographic data for 11 (CCDC 1498511) (CIF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

 Banoub, J.; Boullanger, P.; Lafont, D. Chem. Rev. 1992, 92, 1167.
 Herzner, H.; Reipen, T.; Schultz, M.; Kunz, H. Chem. Rev. 2000, 100, 4495.

- (3) Boltje, T. J.; Buskas, T.; Boons, G. J. Nat. Chem. 2009, 1, 611.
- (4) Sakuda, S.; Isogai, A.; Matsumoto, S.; Suzuki, A. J. Antibiot. 1987, 40, 296.
- (5) Wolfrom, M. L.; Onodera, K. J. Am. Chem. Soc. 1957, 79, 4737.
- (6) Horner, M. W.; Hough, L.; Richards, A. Abstr. Pap. Am. Chem. Soc. 1970, 8.
- (7) James, S. P.; Smith, F.; Stacey, M.; Wiggins, L. F. J. Chem. Soc. 1946, 0, 625.
- (8) Lemieux, R. U.; Ratcliffe, R. M. Can. J. Chem. 1979, 57, 1244.
- (9) Lemieux, R. U.; Nagabhus, T. Can. J. Chem. 1968, 46, 401.
- (10) (a) Emmadi, M.; Kulkarni, S. S. J. Org. Chem. 2011, 76, 4703.
 (b) Emmadi, M.; Kulkarni, S. S. Nat. Protoc. 2013, 8, 1870.
- (11) Kuhn, R.; Fischer, H. Liebigs Ann. Chem. 1958, 617, 88.
- (12) Gigg, R.; Warren, C. D. J. Chem. Soc. 1965, 1351.
- (13) Aguilera, B.; FernandezMayoralas, A. Chem. Commun. 1996, 127.
- (14) Luo, S.-Y.; Kulkarni, S. S.; Chou, C.-H.; Liao, W.-M.; Hung, S.-C. J. Org. Chem. **2006**, *71*, 1226.

(15) Liu, H.-M.; Zhang, F.; Wang, S.-M. Org. Biomol. Chem. 2003, 1, 1641.

- (16) Ko, Y.-C.; Tsai, C.-F.; Wang, C.-C.; Dhurandhare, V. M.; Hu, P.-L.; Su, T.-Y.; Lico, L.-S.; Zulueta, M. M. L.; Hung, S.-C. J. Am. Chem. Soc. 2014, 136, 14425.
- (17) Joseph, A. A.; Dhurandhare, V. M.; Chang, C.-W.; Verma, V. P.; Mishra, G. P.; Ku, C.-C.; Lin, C.-C.; Wang, C.-C. *Chem. Commun.* **2015**, *51*, 104.
- (18) Joseph, A. A.; Verma, V. P.; Liu, X.-Y.; Wu, C.-H.; Dhurandhare, V. M.; Wang, C.-C. *Eur. J. Org. Chem.* **2012**, *2012*, 744.
- (19) Wang, C.-C.; Lee, J.-C.; Luo, S.-Y.; Kulkarni, S. S.; Huang, Y.-W.; Lee, C.-C.; Chang, K.-L.; Hung, S.-C. *Nature* **2007**, *446*, 896.
- (20) Wuts, P. G.; Greene, T. W. Greene's Protective Groups in Organic Synthesis; John Wiley & Sons, 2006.
- (21) Crich, D.; Wu, B. L. Org. Lett. 2006, 8, 4879.
- (22) Volbeda, A. G.; Kistemaker, H. A.; Overkleeft, H. S.; van der Marel, G. A.; Filippov, D. V.; Codee, J. D. C. *J. Org. Chem.* **2015**, *80*, 8796.
- (23) Sanapala, S. R.; Kulkarni, S. S. J. Am. Chem. Soc. 2016, 138, 4938.
- (24) Lee, J.-C.; Tai, C.-A.; Hung, S.-C. Tetrahedron Lett. 2002, 43, 851.
- (25) McGeary, R. P.; Wright, K.; Toth, I. J. Org. Chem. 2001, 66, 5102.

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